



CLEANTECH ALLIANCE  
**BREAKFAST SERIES**  
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# EMERGING CONTAMINANTS PRIMER

CleanTech Alliance, September 9, 2020



## Today's Presenter



### Joseph Quinnan, PE, PG

Sr Vice President/Hydrogeologist

North American Emerging Contaminants Lead

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29 years of experience

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## Today's Presenter



### Erika Houtz, PhD

Senior Environmental Engineer, PFAS Expert

- 12 years of experience
- BS, Chemical Engineer, MS/PhD Environmental Engineering
- Published on the analysis, fate and transport of PFAS in natural and engineered systems, with a particular emphasis on the fate of PFASs found in aqueous film forming foams

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# Emerging Contaminants?

- Impacts to drinking water supplies are driving public awareness
  - Poly and perfluoroalkyl substances (PFAS)
  - 1,4-dioxane
- Changing regulatory standards or debate on toxicity
  - Hexavalent chromium
  - Perchlorate
  - TCE in vapor intrusion
- Evolving understanding of impacts and usage
  - Nanoparticles, pharmaceuticals, pesticides, trichloropropane, flame retardants

# PFAS: Poly and Perfluoroalkyl Substances



# PFAS Introduction



PFAS occur in consumer, industrial, and technical products



Some PFAS are bioaccumulative and potentially toxic; data for most are unavailable



Potential drinking water impacts drive most investigation and treatment efforts



Analytical methods for PFAS are widely available but not always standardized



All PFAS are environmentally stable or form stable end products



PFAS regulatory targets are becoming more widespread and are on a downward trajectory



# PFAS Uses

## Where We Find Them and How They've Evolved



**Firefighting  
Foams**



**Metal  
Plating**



**Textiles**



**Electronics**



**Photography**



**Paper Coatings**



**Paints**



**Hydraulic Fluids**

# Specific Characteristics of PFAS

- **Mobility**  
High aqueous solubility, moderate sorption
- **Extreme Persistence**  
Perfluoroalkyl compounds don't naturally degrade; polyfluorinated compounds form perfluoroalkyl compounds
- **Surfactant Nature**  
Assemble at surfaces, especially air water interfaces
- **Bioaccumulation**  
Long chain PFAS bioaccumulate in humans (protein rich compartments)
- **Toxicity**  
Coupled with bioaccumulation, contributes to low regulations for some PFAS



## **Poly-** and **Per**fluoroalkyl Substances (PFAS)

4,730 known compounds

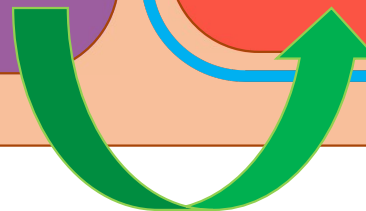
**More Commonly Regulated**

### Polyfluorinated “PFAA Precursors”

1000's of individual parent compounds, and hundreds of common intermediates, e.g. 6:2 FTS, 5:3 acid

### Perfluorinated Compounds(PFCs) or Perfluoroalkyl Acids (PFAAs)

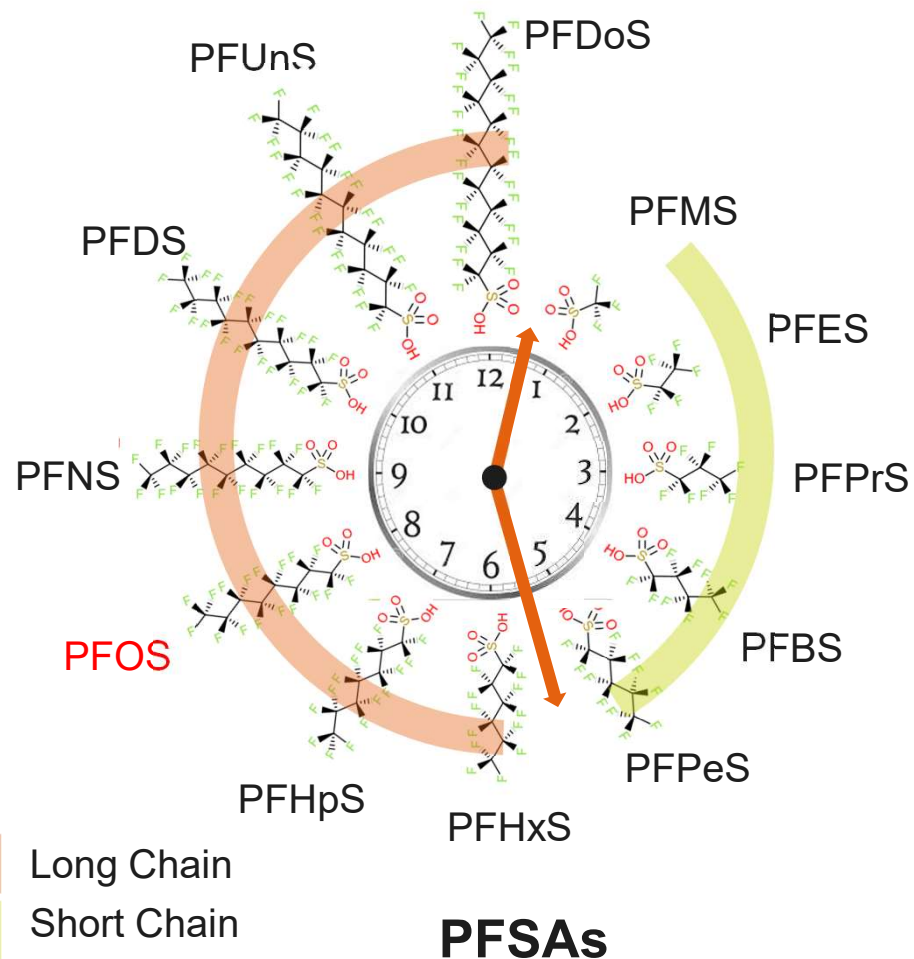
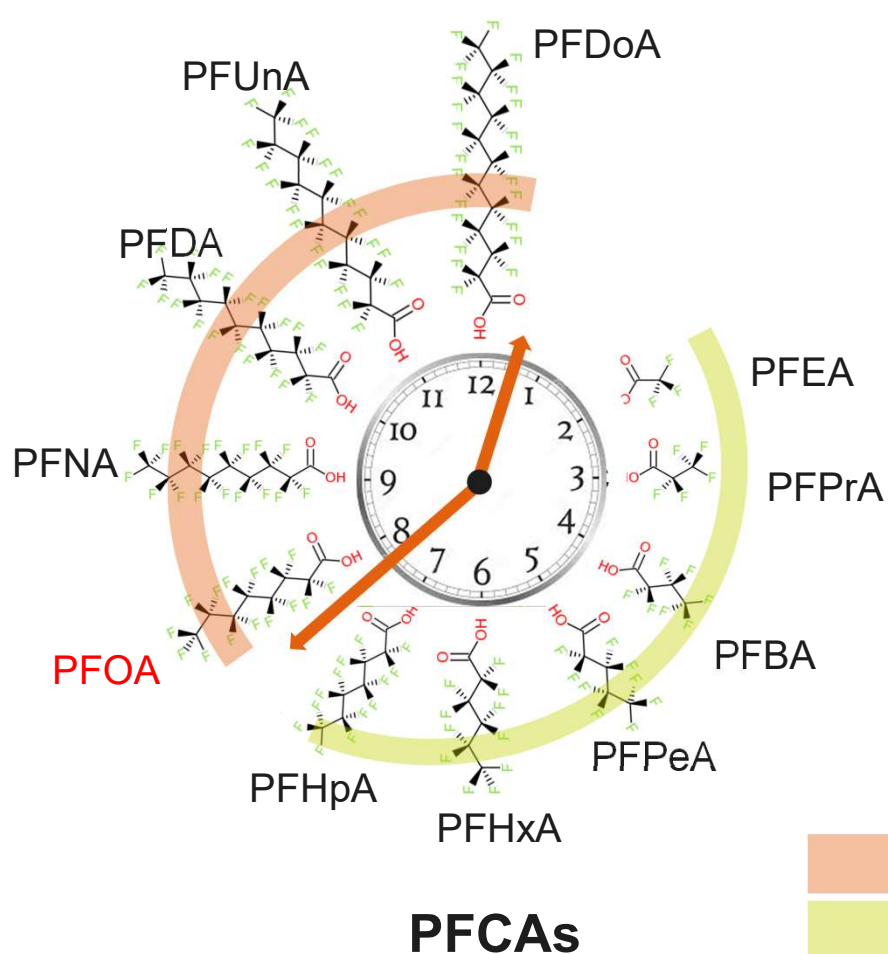
~25 common individual compounds, terminal daughters  
e.g. PFOS, PFOA, PFHxS, PFBA, PFHxA



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**Environmental / Higher Organism Biotransformation**

# Long Chain vs. Short Chain



## NEWS HEADLINES



### The FDA Just Banned These Chemicals in Food. Are They the Tip of the Iceberg?

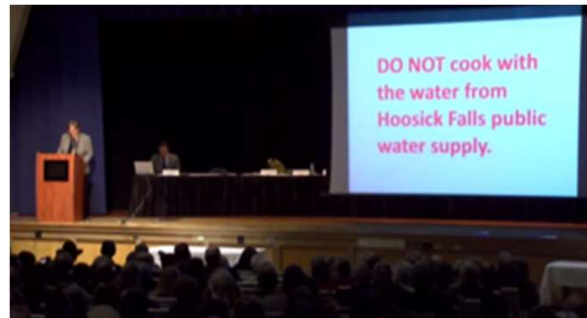
FDA banned three toxic food packaging chemicals and is considering banning seven cancer-causing food flavoring chemicals, but food safety advocates say the process highlights flaws in the system.

By Elizabeth Grossman on January 6, 2016

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### Horsham Water and Sewer Authority Takes 2 Wells Out of Service Due to Detection of Perfluorooctane Sulfonate (PFOS) Above Provisional Health Advisory Level

Two of the Authority supply wells, Well No. 26 and Well No. 40 were recently found to have PFOS above the PHA level (detected at 0.7 ppb and 1.0 ppb respectively). PFOS was also detected in other Authority wells but not above the PHA level. After consulting with DEP, the Authority decided to take Well Nos. 26 and 40 off-line.

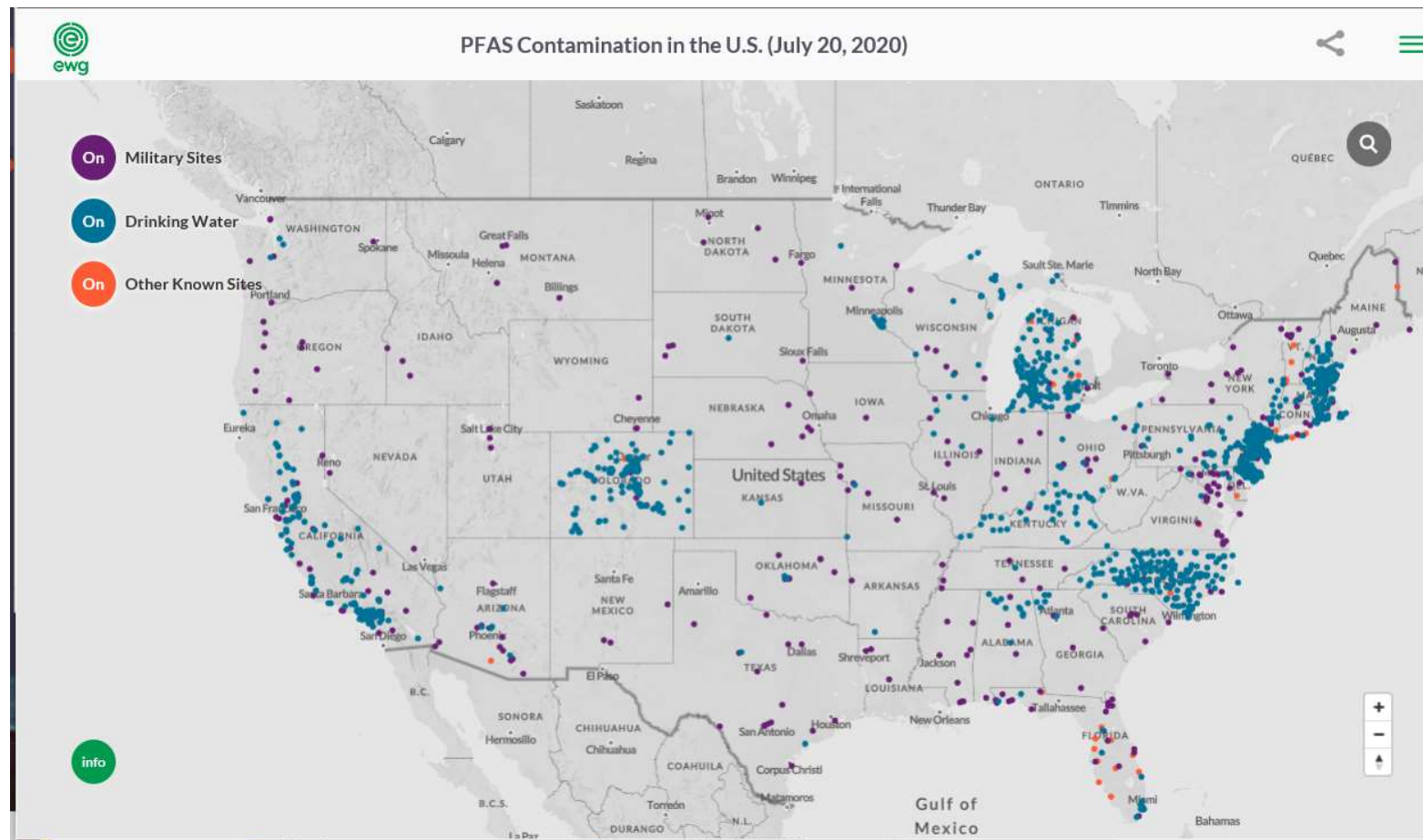


### Toxic chemical foam plume found at National Guard base in Alpena



# PFAS Impacts are widespread

States with  
highest density  
of PFAS sites  
reflects  
programmatic  
testing



## Michigan as Model for Emerging Regulation

- State-wide testing of drinking water (2018)
  - MCLs for PFOS, PFOA and 5 other PFAS
- Wastewater treatment plants (2018)
  - Industrial pre-treatment testing for PFAS users
    - Landfills, Chrome plating, Airports, industry with AFFF fire suppression
- Limits on sewer discharge based on Great Lakes surface water standards
  - Total maximum daily limit of PFAS discharge at each WWTP for permits
- Surface water sampling at industrial sites (2019)
  - Surface water standards among the strictest in country (2018)
  - PFOS 11/12 ng/L, PFOA 420/1200 ng/L for drinking water/non-drinking water sources
- Industrial air emissions standards under development (2020)



## Other Trends Affecting Business

Landfill acceptance of PFAS-containing waste limited in anticipation of hazardous substance regulation

- Subtitle D landfills with specialized leachate treatment
- Zero discharge landfills in arid regions
- Subtitle C landfills

USEPA Significant new use rule (SNUR) and Toxic release Inventory rule (TRIR)

- Increasing restrictions on use and administrative reporting requirements

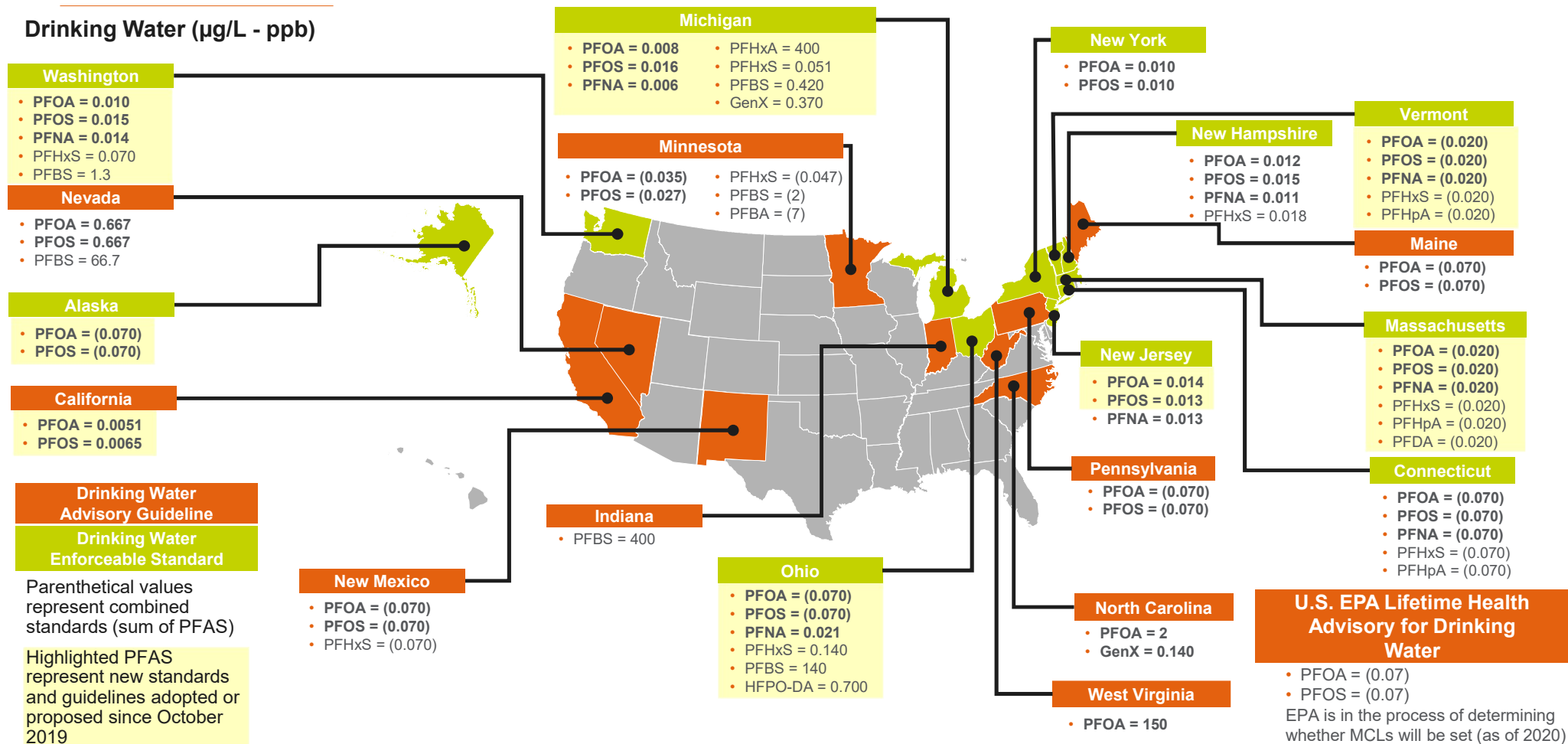
USEPA cancelled incineration testing for PFAS wastes over concerns about incomplete treatment

# Changing U.S. Regulatory Climate

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## Drinking Water (µg/L - ppb)



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Updated August 07, 2020

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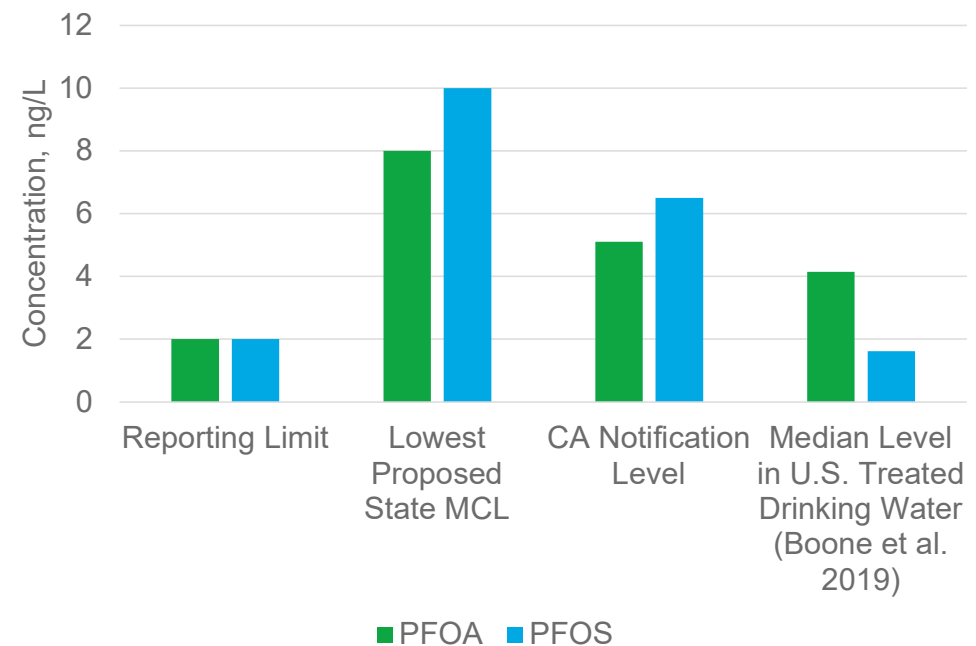
## Reporting Limits and/or regulations are converging on background

Typical reporting limits for PFOS and PFOA in drinking water are 2 ng/L; lower limits are achievable

State-proposed drinking water criteria and enforceable MCLs range between 5 and 20 ng/L

- Recently issued draft CA Environmental Screening Levels are below 1 ng/L

Median levels of PFOS/PFOA are within a factor of ~2x RLs



# PFAS Liquid Treatment Quick Take-Aways

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- PFAS defy conventional remediation engineering
  - does not biodegrade
  - nearly impractical to chemically oxidize
  - has minimal removal through phase changes
  - energy-intensive to destroy
- Current state of the practice is a combination of treatment technologies
- Goal is to concentrate PFAS for energy-intensive destruction

ADSORPTION

SEPARATION/  
CONCENTRATION

DESTRUCTION

# PFAS Solid Treatment Quick Take-Aways

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- Solid treatment technologies available, but **minimal drivers** to necessitate full-scale remediation
- The treatment technologies theoretically/conceptually viable are currently being vetted at the laboratory or pilot scale for effectiveness on short chain PFAS and precursors
- Large mobilization costs complicate small-scale field-scale pilot testing

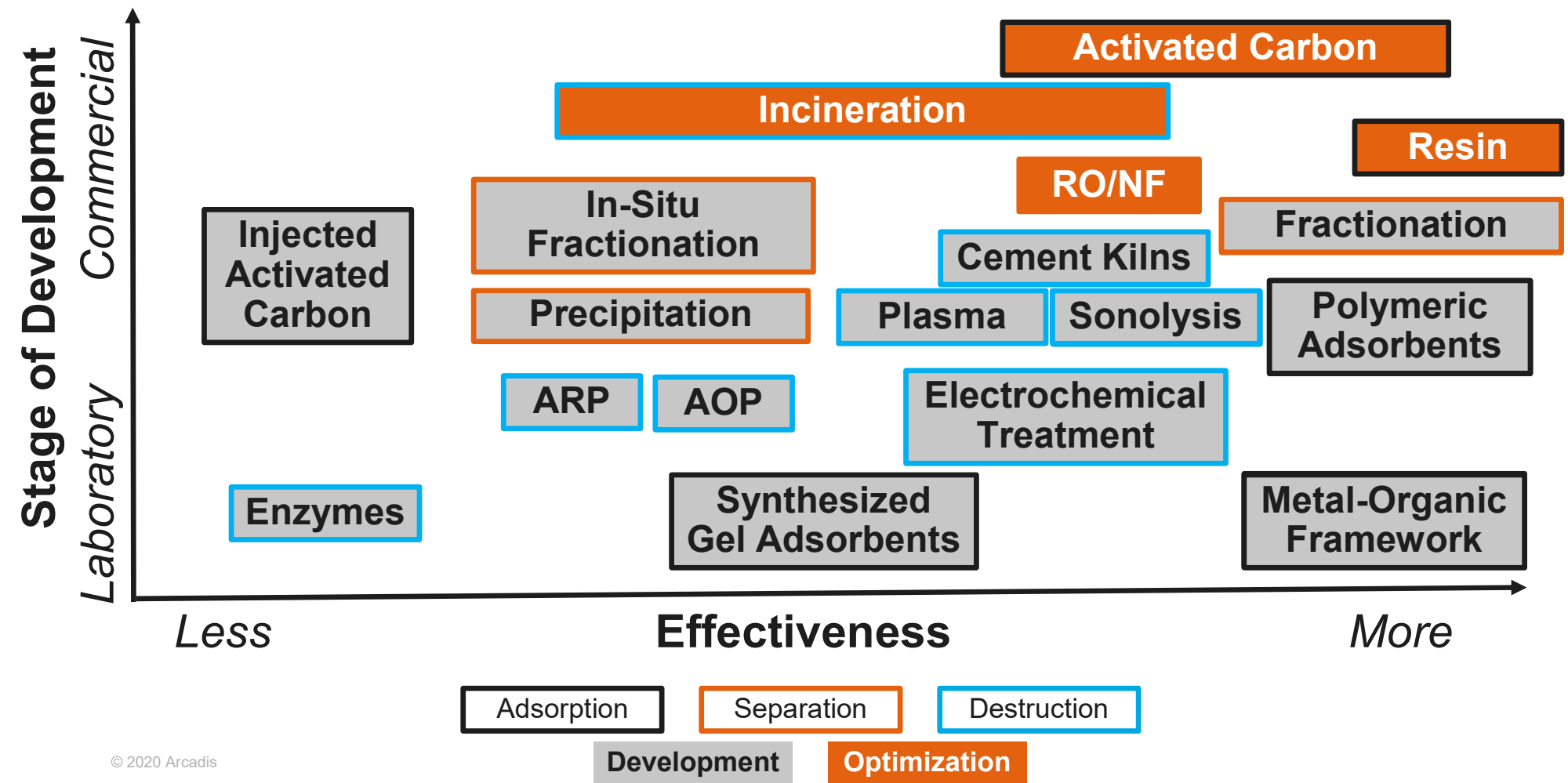
**FIXATION**

**SEPARATION**

**DESTRUCTION**

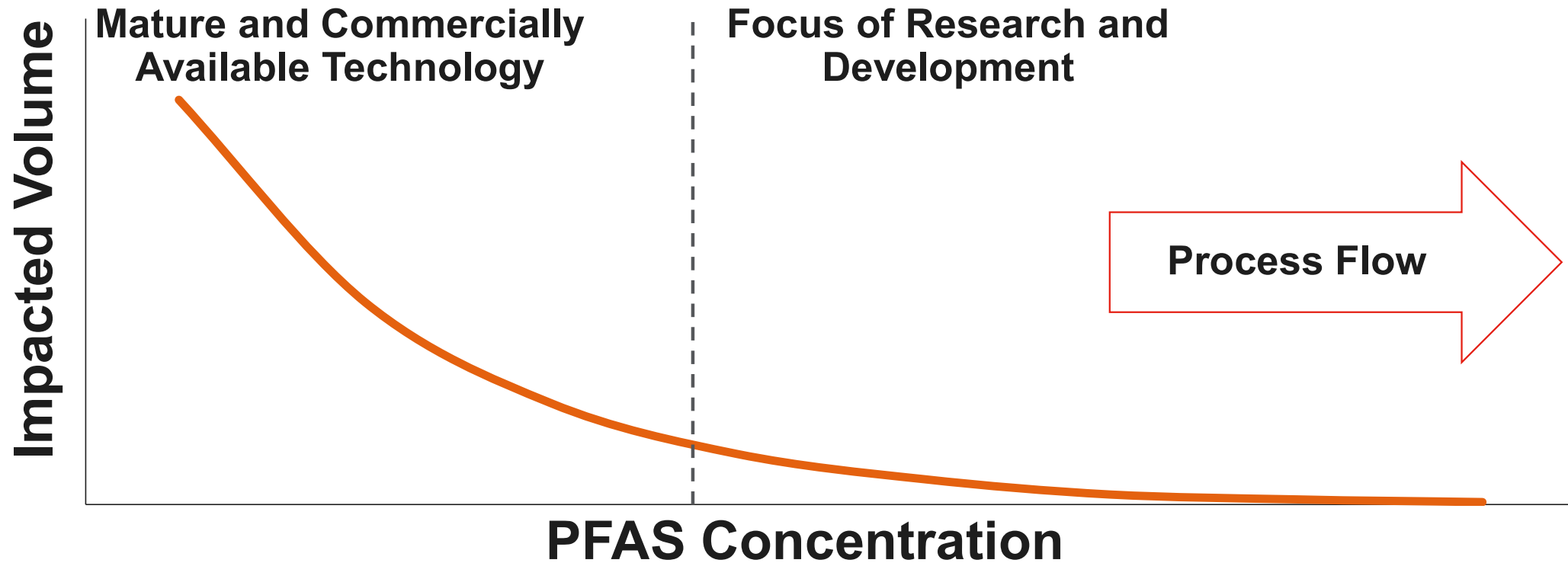
# PFAS Treatment Technologies for Liquid

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# PFAS Treatment State of the Practice

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**Reduce impacted volume while concentrating PFAS for energy-intensive destruction**

# Conventional Technologies for PFAS

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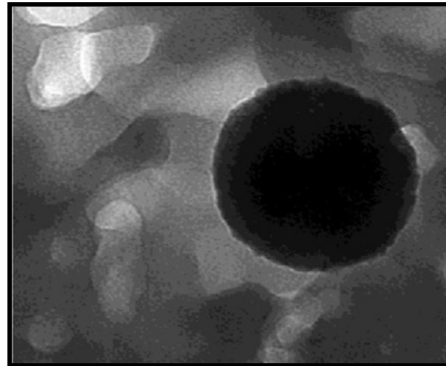


**Activated  
carbon (AC)**



*Photo Source: Evoqua 2017*

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*Photo Source: Zaggia et al. 2016*

**Anion/Ion  
Exchange Resins**



*Photo Source: Evoqua 2017*

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**Reverse  
Osmosis/Nanofiltration**



*Photo Source: Peter Storch 2018*

# PFAS-Relevant Adsorbents

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- GAC/Resins: Current “de facto” IRM adsorbents
- Modified clays (FluoroSorb®), pyrolyzed cellulose, biochar – available, competing with GAC/resin for PFAS relevance
- Cyclodextrin (CycloPure®), Organo-Silica (PQ-Osorb®), customized granular media (Puraffinity®) – promising but experimental
- MOFs, hydrogels, and two-phase composites – somewhat esoteric still, but huge potential adsorption capacities

## Volume of Current Commercial Application

**GAC/Resin**

**Modified Clays, Pyrolyzed Cellulose, Biochar**

**Cyclodextrin, Organo-Silica, custom granular media**

**Metallic Organic Frameworks (MOFs)**

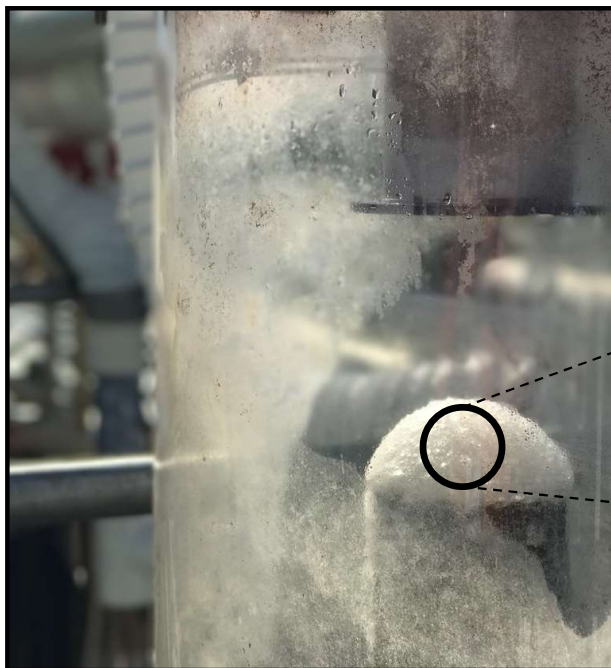


# Fractionation + Ozone = Ozofractionation

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built assets



# Incinerating PFAS in Liquids

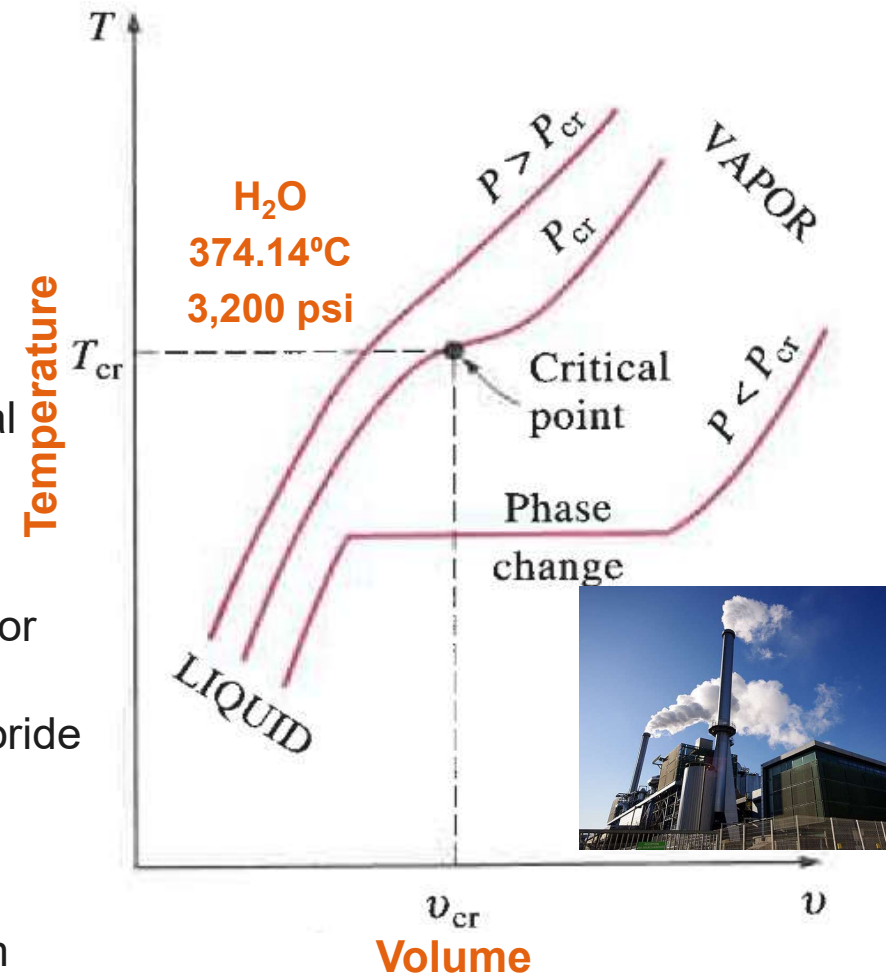
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## Applicability:

- Destruction of PFAS via high temperatures\*
- PFOA defluorination at 300°C to 350°C\*
- PFOS defluorination at 600°C\*.
- Applicable to solids and concentrated waste streams.
- Currently a termination of treatment trains for municipal and small-scale systems.

## Limitations:

- Incomplete mineralization leads to discharge of PFAS or other by-products with long atmospheric half lives\*.
- Complete PFAS mineralization results in hydrogen fluoride (toxic and corrosive)\*.
- Applicability to liquid waste streams may be limited to aqueous critical point\*.
- Insufficient analytical and sampling methods to confirm mineralization



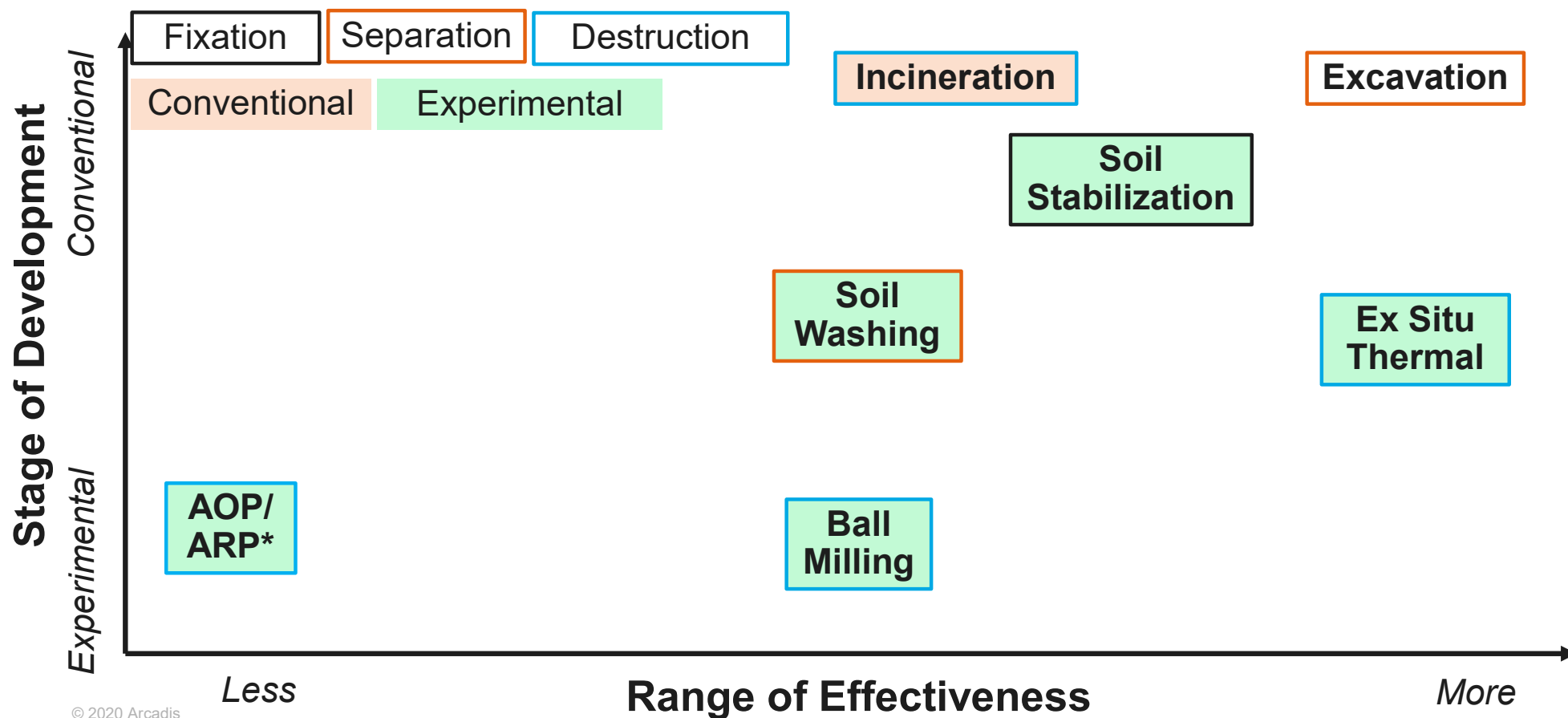
# Advanced Reducing Processes - ARP

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- Thermodynamically possible...
- Kinetically meaningful (scavengers)?

# PFAS Treatment Technologies for Solids

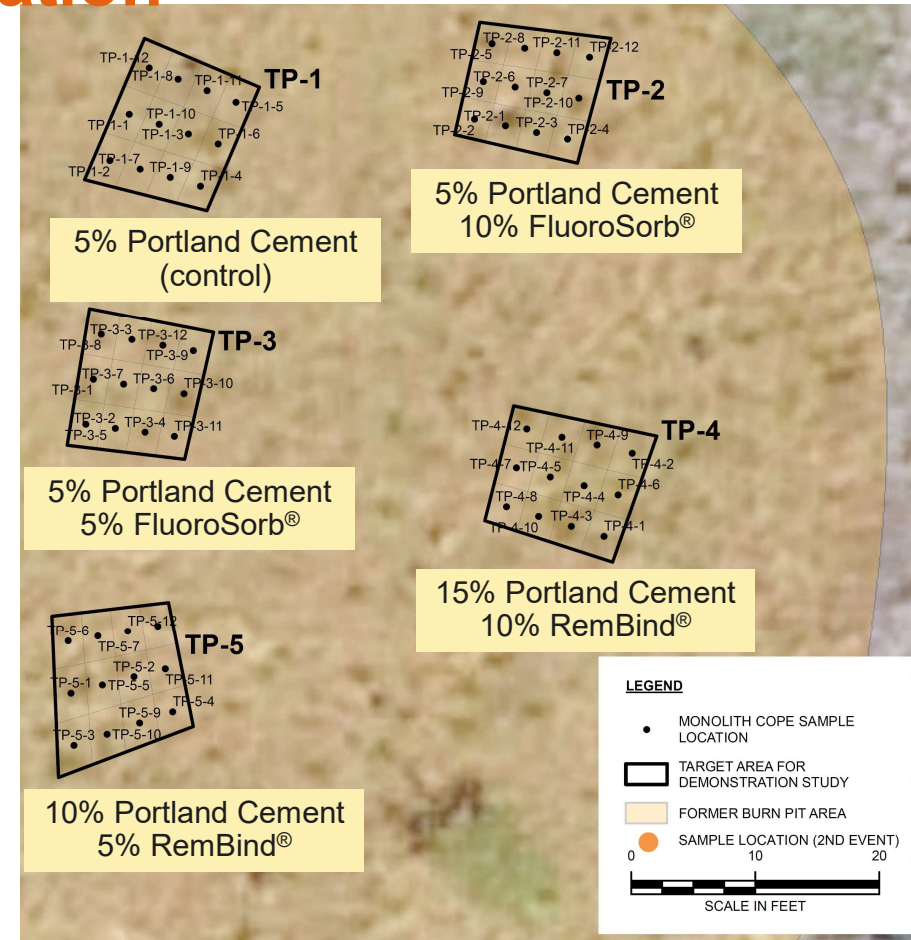
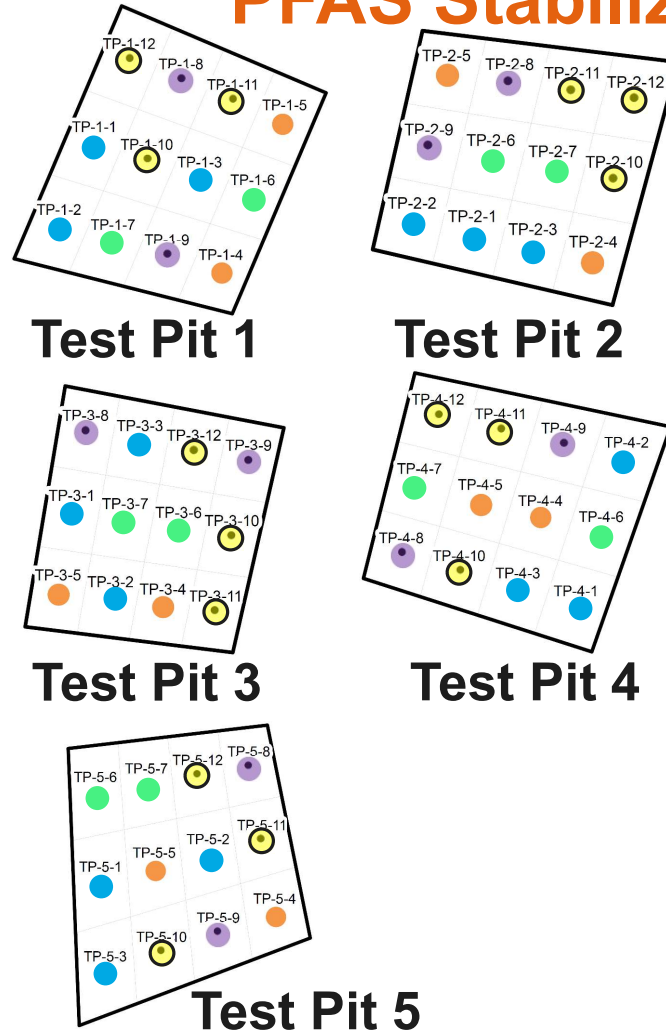




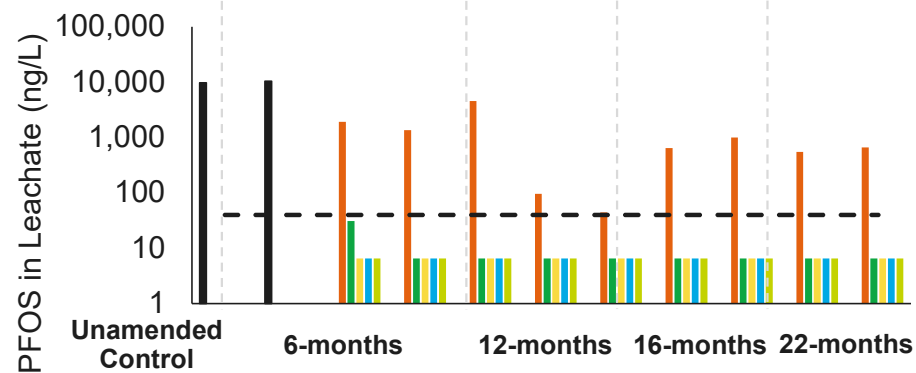
# Field-Scale Demonstration of PFAS Stabilization

- Dec 2018
- Jul 2019
- Oct 2019
- Apr 2020
- \*Oct 2020

\*tentative dates based on access, no dig, etc.



## PFOS



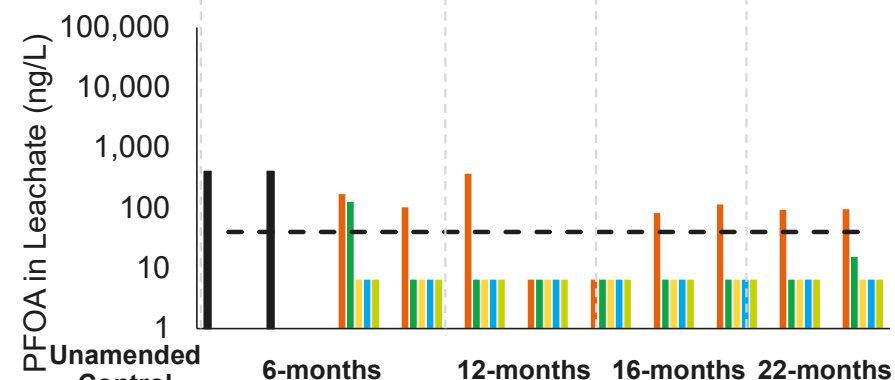
Unamended Control

6-months

12-months

5% FluoroSorb® and  
5% Portland Cement10% FluoroSorb® and  
5% Portland Cement5% RemBind® and  
10% Portland Cement10% RemBind® and  
10% Portland Cement

--- OSD Screening Level



Unamended Control

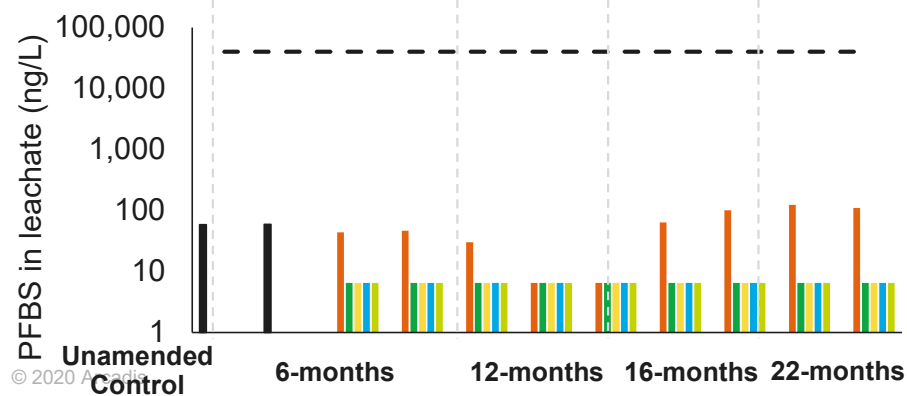
6-months

12-months

16-months

22-months

## PFBS



Unamended Control

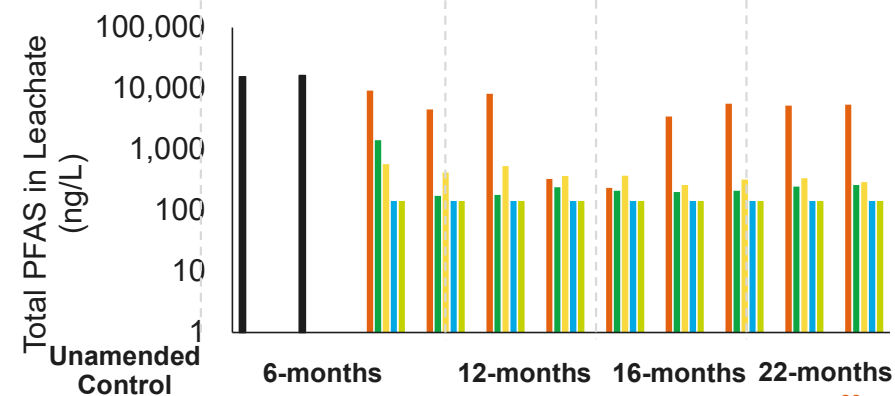
6-months

12-months

16-months

22-months

## Total PFAS - Post TOP Assay



Unamended Control

6-months

12-months

16-months

22-months

# Emerging Contaminants Summary

- Emerging contaminants are considering emerging as the toxicological and regulatory status is evaluated and evolves
- Emerging contaminants such as 1,4-dioxane and PFAS present a new set of challenges to practitioners
- In situ and ex situ management strategies are being developed, remedial technologies in research and development
- Practical laboratory quantification is a topic of focus as emerging contaminants have low targets



# Q&A

# Thank you!

Contact us:

Joe Quinnan

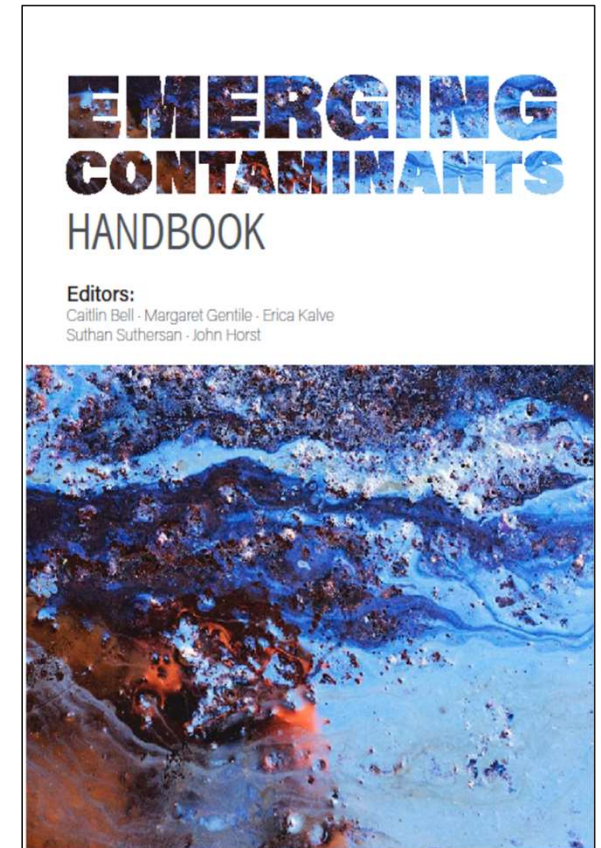
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CRC Press 2019

Download  
**PFAS in  
Perspective**  
for a closer look at  
how different  
stakeholders are  
approaching PFAS.



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